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Textbookshut

In chemistry, resonance or mesomerism is a way of describing delocalized electrons within certain molecules or polyatomic ions where the bonding cannot be expressed by one single Lewis structure. A molecule or ion with such delocalized electrons is represented by several contributing structures (or forms) (also variously known as resonance structures (or forms), canonical structures, or, in ...

Resonance (chemistry) - Wikipedia

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Isomerases catalyze changes within one molecule. They convert one isomer to another, meaning that the end product has the same molecular formula but a different physical structure.

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